

## Functionalized magnetite / silica nanocomposite for oily wastewater treatment

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**Abstract.** A new magnetite-silica core/shell nanocomposite ( $\text{Fe}_3\text{O}_4@\text{nSiO}_2@\text{mSiO}_2$ ) was synthesized and functionalized with trimethylchlorosilane (TMCS). The prepared nanocomposite was used for the removal of diesel oil from aqueous media. The characterization of magnetite-silica nanocomposite was studied by X-ray diffraction (XRD), Fourier transform infrared (FTIR), transmission electron microscopy (TEM), surface area measurement, and vibrating sample magnetization (VSM). Results have shown that the desired structure was obtained and surface modification was successfully carried out. FTIR analysis has confirmed the presence of TMCS on the surface of magnetite silica nanocomposites. The low-angle XRD pattern of nanocomposites indicated the mesoscopic structure of silica shell. Furthermore, TEM results have shown the core/shell structure with porous silica shell. Adsorption kinetic studies indicated that the nanocomposite was able to remove 80% of the oil contaminant during 2 h and fit well with the pseudo-second order model. Equilibrium studies at room temperature showed that the experimental data fitted well with Freundlich isotherm. The magnetic property of nanocomposite facilitated the separation of solid phase from aqueous solution.

**Keywords:** magnetite silica nanocomposite; core-shell; oily pollutant; adsorption; functionalized surface

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### 1. Introduction

Over recent years most waters have become increasingly polluted by oil. A large amount of oil spills into the aquatic ecosystem can cause serious environmental problems, including clogging of sewage treatment plants, adversely affecting the aquatic biota, and increasing biochemical oxygen demand due to large amount of bacteria necessary to decompose the oil (Wang *et al.* 2010a). The most frequent oils that are found in oily wastewater include lubricants, cutting oil, vegetable oil, light and heavy hydrocarbons. Major industrial sources of oily wastewater are petroleum refining and petrochemical plants, steel manufacturing and metal working, vehicle repair (Wang *et al.* 2012), and food processing.

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- Kitis, M., Harman, B.I., Yigit, N.O., Beyhan, M., Nguyen, H. and Adams, B. (2007), "The removal of natural organic matter from selected Turkish source waters using magnetic ion exchange resin (MIEX®)", *React. Funct. Polym.*, **67**(12), 1495-1504.
- Largergren, S. (1898), "Zur theorie der sogenannten adsorption geloster stoffe. Kungliga Svenska Vetenskapsakademiens", *Handlingar*, **24**, 1-39.
- Liu, Z., Yang, H., Zhang, H., Huang, C. and Li, L. (2012), "Oil-field wastewater purification by magnetic separation technique using a novel magnetic nanoparticle", *Cryogenics*, **52**(12), 699-703.
- Mahmoud, M.E., Abdelwahab, M.S. and Fathallah, E.M. (2013), "Design of novel nano-sorbents based on nano-magnetic iron oxide-bound-nano-silicon oxide-immobilized-triethylenetetramine for implementation in water treatment of heavy metals", *Chem. Eng. J.*, **223**, 318-327.
- Niu, H.-Y., Li, W.-H., Shi, Y.-L. and Cai, Y.-Q. (2011), "A core-shell magnetic mesoporous silica sorbent for organic targets with high extraction performance and anti-interference ability", *Chem. Commun.*, **47**(15), 4454-4456.
- Oh, J.K. and Park, J.M. (2011), "Iron oxide-based superparamagnetic polymeric nanomaterials: design, preparation, and biomedical application", *Prog. Polym. Sci.*, **36**(1), 168-189.
- Oliveira, L.C., Petkowicz, D.I., Smaniotto, A. and Pergher, S.B. (2004), "Magnetic zeolites: a new adsorbent for removal of metallic contaminants from water", *Water Res.*, **38**(17), 3699-3704.
- Rorrer, G.L., Hsien, T.Y. and Way, J.D. (1993), "Synthesis of porous-magnetic chitosan beads for removal of cadmium ions from wastewater", *Ind. Eng. Chem. Res.*, **32**(9), 2170-2178.
- Savina, I.N., English, C.J., Whitby, R.L., Zheng, Y., Leistner, A., Mikhalovsky, S.V. and Cundy, A.B. (2011), "High efficiency removal of dissolved As (III) using iron nanoparticle-embedded macroporous polymer composites", *J. Hazard. Mater.*, **192**(3), 1002-1008.
- Srinivasan, A. and Viraraghavan, T. (2008), "Removal of oil by walnut shell media", *Bioresour. Technol.*, **99**(17), 8217-8220.
- Syamimi Zaidi, N., Sohaili, J., Muda, K. and Sillanpää, M. (2014), "Magnetic field application and its potential in water and wastewater treatment systems", *Sep. Purif. Rev.*, **43**(3), 206-240.
- Wang, D., Silbaugh, T., Pfeffer, R. and Lin, Y. (2010a), "Removal of emulsified oil from water by inverse fluidization of hydrophobic aerogels", *Powder Technol.*, **203**(2), 298-309.
- Wang, J., Zheng, S., Shao, Y., Liu, J., Xu, Z. and Zhu, D. (2010b), "Amino-functionalized Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub> core-shell magnetic nanomaterial as a novel adsorbent for aqueous heavy metals removal", *J. Colloid Interface Sci.*, **349**(1), 293-299.
- Wang, D., McLaughlin, E., Pfeffer, R. and Lin, Y. (2012), "Adsorption of oils from pure liquid and oil-water emulsion on hydrophobic silica aerogels", *Sep. Purif. Technol.*, **99**, 28-35.
- Wu, F.-C., Tseng, R.-L. and Juang, R.-S. (2000), "Comparative adsorption of metal and dye on flake-and bead-types of chitosans prepared from fishery wastes", *J. Hazard. Mater.*, **73**(1), 63-75.
- Xu, X., Deng, C., Gao, M., Yu, W., Yang, P. and Zhang, X. (2006), "Synthesis of magnetic microspheres with immobilized metal ions for enrichment and direct determination of phosphopeptides by matrix-assisted laser desorption ionization mass spectrometry", *Adv. Mater.*, **18**(24), 3289-3293.
- Xu, Y., Zhou, Y., Ma, W., Wang, S. and Li, S. (2013), "Functionalized magnetic core-shell Fe<sub>3</sub>O<sub>4</sub>@SiO<sub>2</sub> nanoparticles for sensitive detection and removal of Hg<sup>2+</sup>", *J. Nanopart. Res.*, **15**(6), 1-9.
- Zhang, G., Qu, J., Liu, H., Cooper, A.T. and Wu, R. (2007), "CuFe<sub>2</sub>O<sub>4</sub>/activated carbon composite: a novel magnetic adsorbent for the removal of acid orange II and catalytic regeneration", *Chemosphere*, **68**(6), 1058-1066.
- Zhao, X., Shi, Y., Wang, T., Cai, Y. and Jiang, G. (2008), "Preparation of silica-magnetite nanoparticle mixed hemimicelle sorbents for extraction of several typical phenolic compounds from environmental water samples", *J. Chromatogr. A*, **1188**(2), 140-147.
- Zhong, L.-S., Hu, J.-S., Cao, A.-M., Liu, Q., Song, W.-G. and Wan, L.-J. (2007), "3D flowerlike ceria micro/nanocomposite structure and its application for water treatment and CO removal", *Chem. Mater.*, **19**(7), 1648-1655.