

Response surface analysis of removal of a textile dye by a Turkish coal powder

Alireza Khataee^{*1}, Leila Alidokht^{1,2}, Aydin Hassani³ and Semra Karaca³

¹ Research Laboratory of Advanced Water and Wastewater Treatment Processes, Department of Applied Chemistry, Faculty of Chemistry, University of Tabriz, Tabriz, Iran

² Department of Soil Science, Faculty of Agriculture, University of Tabriz, Tabriz, Iran

³ Department of Chemistry, Faculty of Science, Atatürk University, 25240 Erzurum, Turkey

(Received November 08, 2013, Revised December 19, 2013, Accepted December 20, 2013)

Abstract. In the present study, an experimental design methodology was used to optimize the adsorptive removal of Basic Yellow 13 (BY13) using Turkish coal powder. A central composite design (CCD) consisting of 31 experiments was employed to evaluate the simple and combined effects of the four independent variables, initial dye concentration (mg/L), adsorbent dosage (g/L), temperature (°C) and contact time (min) on the color removal (CR) efficiency (%) and optimizing the process response. Analysis of variance (ANOVA) showed a high coefficient of determination value ($R^2 = 0.947$) and satisfactory prediction of the polynomial regression model was derived. Results indicated that the CR efficiency was not significantly affected by temperature in the range of 12-60°C. While all other variables significantly influenced response. The highest CR (95.14%), estimated by multivariate experimental design, was found at the optimal experimental conditions of initial dye concentration 30 mg/L, adsorbent dosage 1.5 g/L, temperature 25°C and contact time 10 min.

Keywords: adsorption; coal; organic dye; experimental design; optimization

1. Introduction

The dye effluents are considered to be harmful to aquatic environments and interfere with light penetration in the receiving water bodies which ultimately disturb the biological processes (Garg *et al.* 2004). To treat dye-containing effluents, several physical and chemical processes, such as coagulation/flocculation, biosorption, photocatalytic degradation, ultrafiltration and advanced oxidation processes (AOPs) have been applied (Sadri Moghaddam *et al.* 2010, Kousha *et al.* 2012, Mozia *et al.* 2008, Dong *et al.* 2011, Khataee *et al.* 2011, Modirshahla *et al.* 2011, Chen *et al.* 2005). The literature indicates that in the recent years adsorption techniques have received much attention for this purpose, offering significant reduction of expenses and efficient removal of dyes. Since the adsorption of dye by an adsorbent is strongly influenced by many factors, including adsorbent dosage, initial effluent pH, initial concentration of dye, temperature and the contact time of adsorbent with dye, it is crucial to search for the key influencing factor(s) and discover the experimental conditions in which the best possible response can be obtained. Performing such

*Corresponding author, Professor, E-mail: a_khataee@tabrizu.ac.ir; ar_khataee@yahoo.com

- water by magnetic nanocomposite using response surface modeling approach”, *J. Hazard. Mater.*, **186**(2-3), 1462-1473.
- Speight, J.G. (1972), “The application of spectroscopic techniques to the structural analysis of coal and petroleum”, *Appl. Spectrosc. Rev.*, **5**(1), 211-263.
- Tanyildizi, M.Ş. (2011), “Modeling of adsorption isotherms and kinetics of reactive dye from aqueous solution by peanut hull”, *Chem. Eng. J.*, **168**(3), 1234-1240.
- Tseng, R.-L. and Tseng, S.-K. (2006), “Characterization and use of high surface area activated carbons prepared from cane pith for liquid-phase adsorption”, *J. Hazard. Mater.*, **136**(3), 671-680.
- Wu, S., Yu, X., Hu, Z., Zhang, L. and Chen, J. (2009), “Optimizing aerobic biodegradation of dichloromethane using response surface methodology”, *J. Environ. Sci.*, **21**(9), 1276-1283.

CC